# A NEW ANALYTICAL METHOD DEVELOPMENT AND VALIDATION OF PIPIERAQUINE PHOSPHATE AND ARTEROLANE MALEATE BY RP-HPLC METHOD <br> Thandava Krishna V *, Aneesha Rani, Nandha Kishore A, S. Duraivel <br> Nimra college of Pharmacy, Jupudi, Vijayawada <br> *Corresponding author: E-mail- krishnareddyvaddiboina@gmail.com 


#### Abstract

A simple, economical, precise, accurate and rapid HPLC method has been developed and validated for assay determination of Piperaquine Phosphate and Arterolane maleate simultaneously in their tablet dosage forms. The chromatographic conditions was performed using Waters Hypersil ODS ( $150 * 4.6 \mathrm{~mm} * 5 \mu$ ) Column on a mixture of Methanol: 0.01 M Potassium dihydrogen phosphate ( pH 2.6 ) in the ratio of $(60: 40 \% \mathrm{v} / \mathrm{v})$. The detection was carried out at 240 nm with the flow rate of $1.0 \mathrm{ml} / \mathrm{min}$. The retention times for $\mathrm{PQP}, \mathrm{AM}$ were 3.39 and 5.53 min respectively. This method shows to be Linear ( $\mathrm{R}^{2}>0.99$ ), precise ( $\mathrm{RSD}<2 \%$ ), Accurate Recovery of $(98-102 \%$ ) of PQP and AM. The proposed method was successfully employed for the drug contents in marketed formulations, according to ICH guidelines and found to be suitable for the simultaneous estimation of the drugs.


Key words: Piperaquine Phosphate, Arterolane maleate, RP-HPLC, Combined dosage form.

## 1. INTRODUCTION

Piperaquine Phosphate is chemically known as 7-chloro-4-[4-[3-[4-(7- Chloroquinolin-4-yl) piperazin-1yl]prppyl] piperazin-1-yl]quinolne; phosphoric acid. Arterolane maleate is chemically known as [( N -(2-amino-2-methyl propyl)-2-cis-(adamantine-2, ${ }^{\prime}$ '-[1,2,4] trioxolane-5', 1 "-cyclohexan)-4"-yl] acetamide: maleate (Tripathi KD.,2008). PQP and AM acts as Anti-Malarial drugs. PQP interferes with the degradation of haemoglobin by parasitic lysosomes, helps in damaging of plasmodial membranes. AM causes lipid peroxidation, damages endoplasmic reticulum, inhibits protein synthesis and ultimately results in lysis of the parasite ${ }^{3}$. Highly sensitive, selective, HPLC method will be very useful for the estimation of PQP and AM in combined dosage formulations. Literature survey reveals need for simultaneous estimation of drugs. Few methods were reported by HPLC, LC-MS, Capillary zone electroporosis. The purpose of this study was to develop sensitive, simple, precise, accurate method for simultaneous estimation of PQP and AM in bulk and combined dosage form (Debrus B., 2011).



Figure.1.Chemical Structure of Piperaquine phosphate Figure.2.Chemical Structure of Arterolane maleate MATERIALS AND METHODS
Apparatus: Separation and estimation was carried out using HPLC (waters- 2469 with PDA detector), column used in experiment was $\mathrm{C}_{18}$ Waters Hypersil ODS, $5 \mu\left(150^{*} 4.6 \mathrm{~mm}\right)$ analytical balance used was LAB INDIA, Digital pH meter LAB INDIA. The mobile phase was prepared by mixing 0.01 M Potassium dihydrogen phosphate buffer: Methanol $(\mathrm{pH}-$ 2.6 ) in the ratio of ( $60: 40$ ) was filtered and degassed. Injection volume is $10 \mu \mathrm{~L}$ and the detection was at 240 nm .

Reagents and solutions: Pure sample of Piperaquine phosphate and Arterolane maleate and other reagents such as Methanol, milliQ water of HPLC grade and Potassium dihydrogen phosphate.

Preparation of mobile phase: The mobile phase was prepared by mixing 0.01 M Potassium dihydrogen phosphate: Methanol ( $\mathrm{pH}-2.6$ ) in the ratio of ( $60: 40 \% \mathrm{v} / \mathrm{v}$ ) and was filtered and degassed.

Preparation of standard drug solution: About 15.0 mg of $\mathrm{AM}, 75.0 \mathrm{mg}$ of PQP standard were weighed accurately into 100 ml of volumetric flask. Add 70 ml of Diluent was added and sonicated for 5 minutes. After sonication, the volume was made up to the mark with diluent to obtain final concentration of $150 \mu \mathrm{~g} / \mathrm{ml}, 750 \mu \mathrm{~g} / \mathrm{ml}$ AM and PQP respectively. (Wahajuddin., 2013) (Tarning J., 2006).
Marketed Formulation: 20 tablets were accurately weighed and average weight was determined. Powdered tablet equivalent to 150 mg of AM and 750 mg of PQP was transferred into a 100 ml volumetric flask. 70 ml of water was added and sonicated for 10 minutes. The volume was made upto mark with diluents after cooling and mixed well, filtered through $0.45 \mu$ nylon filter. Finally get concentration of AM as about $1500 \mu \mathrm{~g} / \mathrm{ml}$, PQP $7500 \mu \mathrm{~g} / \mathrm{ml}$. The solution was
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filtered through $0.45 \mu$ nylon filter. Further 5 ml from this solution was diluted to 50 ml with diluent and mixed well to get concentration of $150 \mu \mathrm{~g} / \mathrm{ml}$ of AM, $750 \mu \mathrm{~g} / \mathrm{ml}$ of PQP. (Hodel E., 2009) (Kirchhofer C., 2010)

Table.1.Details of marketed Formulation

| Brand Name | Content | Mfg. Company |
| :--- | :--- | :--- |
| Synriam | Piperaquine Phosphate\& Arterolane <br> maleate(750 mg \& 150mg Respectively) | Ranbaxy |

Selection of mobile phase for method Optimization and experimental condition: Several trial had been taken for the proper optimization of HPLC method by changing different mobile phase with different ratio. And finally the mobile phase for optimised condition 0.01 M potassium dihydrogen phosphate: Methanol ( $\mathrm{pH}-2.6$ ) in the ratio of $(60: 40)$ was selected and chromatogram was shown Fig no.3.


Figure.3.Chromatogram of PQP and AM for optimised method
Table: 2 Chromatographic conditions for the optimized method

| Parameters | Description |
| :--- | :--- |
| Column | Waters Hypersil ODS C18 $(150 * 4.6 \mathrm{~mm}, 5.0 \mu)$ |
| Mobile phase | 0.01 M Potassium dihydrogen Phosphate $:$ Methanol $(60: 40)$ |
| Injection volume | $10 \mu \mathrm{l}$ |
| Flow rate | $1.0 \mathrm{ml} / \mathrm{min}$ |
| Detection wavelength | 240 nm |
| Column Temparature | $40^{\circ} \mathrm{C}$ |
| Sample Temparature | $25^{\circ} \mathrm{c}$ |
| Run Time | 10 min |

Assay: Assay of marketed formulation containing PQP ( 750 mg ) and AM ( 150 mg ) was performed by preparing the sample solution as describer earlier in the preparation of the sample. The assay of the commercial sample was calculated by comparing the area of standard and sample peaks. The assay of marketed formulation synriam was found to be within the limit; the chromatogram is shown in Fig 4.


Figure.4.Chromatogram for the Assay of marketed Formulation
This optimized method was validated terms of Linearity, accuracy, Precision, Specificity, as per ICH guidelines.

Linearity: The linearity of calibration curves (peak area $\mathrm{V}_{\mathrm{s}}$ concentration) in pure solution was checked over the concentration ranges of about $50-150 \%$ (Assay concentration $\mu \mathrm{g} / \mathrm{ml}$ ) for PQP and AM. The total eluting time was less than 10.0 min . The calibration curve were linear in the studied range and equations of the regression analysis were obtained $y=-0.1152 x, R^{2}=0.999$ for PQP and $y=-0.7204 . x, R^{2}=0.999$ for $A M$.
Accuracy: Accuracy of the method was determined by recovery experiments at spiked levels of $50 \%, 100 \%, 150 \%$. The recovery studies were carried out three times, the percentage recovery and percentage relative standard deviations were calculated.

Precision: The precision of the analytical method was studied by analysis of multiple sampling of same homogeneous sample.
Intra-day Precision: Intra-day precision was determined by analyzing the combined standard solutions of AM and PQP $(150,750 \mu \mathrm{~g} / \mathrm{ml})$ at three different time intervals on same day.
Inter-day Precision: Inter-day precision was determined by analyzing the combined standard solutions of AM and PQP ( $150,750 \mu \mathrm{~g} / \mathrm{ml}$ ) on three consecutive days.
Specificity: A solution containing a mixture of tablet was prepared using sample preparation procedure and injected in to the system, to evaluate possible interfering peaks.
Forced degradation: Forced degradation studies were carried out on the sample preparations of AM, PQP tablets $150: 750 \mathrm{mg}$ and the degradation was evaluated by calculating the $\%$ degradation of AM and PQP in comparison with unstressed sample preparation. It was tried to achieve degradation of these drug between $10 \%$ and $30 \%$ by following stress conditions to prove the method as a stability indicating method.
The following are the stress conditions which were followed for forced degradation studies;
Acid Degradation: Weigh accurately about 1368 mg of Sample in 100 ml volumetric flask and 20 ml of diluents and add 5 ml of 5 N HCl boil for $1 \mathrm{hr}\left(80^{\circ} \mathrm{c}\right)$ and neutralize with 5 N NaOH and make up to the mark. 5 ml of this solution is diluted in to 50 ml and filtered through the 0.45 nylon filtered and analyse the recorded chromatogram.


Fig No.5:Chromatogram of Combined Tablet Solution of PQP, AM in Acid Stress Degradation


Figure.6.Chromatogram of Purity Plot of PQP and AM in Acid Stress Degradation
www.jchps.com Journal of Chemical and Pharmaceutical Sciences Alkali Degradation: Weigh accurately about 1368 mg of Sample in 100 ml volumetric flask and 20 ml of diluents and add 5 ml of 5 N NaOH boil for $1 \mathrm{hr}\left(80^{\circ} \mathrm{c}\right)$ and neutralize with 5 N HCl and make up to the mark. 5 ml of this solution is diluted in to 50 ml and filtered through the 0.45 nylon filtered and analyse the recorded chromatogram.


Figure.7.Chromatogram of Combined Tablet Solution of PQP and AM in Alkali Stress Degradation


Figure.8.Chromatogram of Purity Plot of PQP \& AM in Alkali Stress Degradation
Peroxide Degradation: Weigh accurately about 1368 mg of Sample in 100 ml volumetric flask and 20 ml of diluents and add 5 ml of $6 \% \mathrm{H}_{2} \mathrm{O}_{2}$ boil for 1 hr (800c) and make up to the mark. 5 ml of this solution is diluted in to 50 ml and filtered through the 0.45 nylon filtered and analyse the recorded chromatogram.


Figure.9.Chromatogram of Combined Tablet Solution of PQP and AM in Peroxide Stress Degradation


Figure.10.Chromatogram of Purity Plot of PQP in Peroxide Stress Degradation
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## RESULTS AND DISCUSSION

## Validation:

Table.3.Method Precision Results for Piperaquine Phosphate

| Test No. | Trial 1 | Trial 2 | Trial 3 | Trial 4 | Trial 5 | Trial 6 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Avg.Wt. (mg) | $\mathbf{1 3 6 8 . 2 3}$ |  |  |  |  |  |
| Wt. taken (mg) | 1368.01 | 1359.98 | 1367.98 | 1368.26 | 1368.00 | 1368.35 |
| Area (Inj.1) | 8563656 | 8568156 | 8571651 | 8566376 | 8563546 | 8565654 |
| Area (Inj.2) | 8563685 | 8565658 | 8553658 | 8566686 | 8559696 | 8561686 |
| Avg. Area | 8563670 | 8563457 | 8563465 | 8566321 | 8561236 | 8562315 |
| \% RSD | 0.1 | 0.2 | 0.1 | 0.2 | 1.1 | 0.1 |
| Assay(mg/tab) | 750.52 | 750.70 | 751.06 | 750.09 | 750.92 | 750.62 |
| Assay (\%) | 100.1 | 101.2 | 101.8 | 100.2 | 101.9 | 100.9 |
| Average Assay | In mg= 750.65mg | In \%=100.92\% |  |  |  |  |
| STDV | For mg assay=0.6141 | For \% assay=0.4262 |  |  |  |  |
| \%RSD | For mg assay=0.4 | For \% assay=0.4 |  |  |  |  |

Table.4. Method Precision Results for Arterolane maleate

| Test No. | Trial 1 | Trial 2 |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Trial 3 |  |  |  |  |  | Trial 4 |
| Avg.Wt. (mg) | $\mathbf{1 3 6 9 . 0 5}$ | Trial 5 | Trial 6 |  |  |  |
| Wt.taken( mg) | 1368.75 | 1369.25 | 1368.95 | 1368.05 | 1367.95 | 1368.55 |
| Area (Inj.1) | 1132502 | 1132512 | 1132562 | 1140118 | 1132758 | 1132856 |
| Area (Inj.2) | 1132635 | 1140212 | 1136574 | 1139925 | 1135555 | 1132641 |
| Avg. Area | 1132569 | 1136362 | 1134568 | 1140022 | 1134157 | 1132749 |
| \% RSD | 0.2 | 0.1 | 0.0 | 0.2 | 1.1 | 0.1 |
| Assay(mg/tab) | 151.06 | 151.15 | 151.09 | 152.03 | 151.09 | 151.05 |
| Assay (\%) | 100.9 | 101.2 | 101.0 | 101.9 | 101.0 | 100.8 |
| Average Assay | In $\mathrm{mg}=151.28 \mathrm{mg}$ | In \% $=101.4 \%$ |  |  |  |  |
| STDV | For mg assay=1.3058 | For \% assay=0.4243 |  |  |  |  |
| \%RSD | For mg assay=0.4 | For \% assay=0.4 |  |  |  |  |

Table.5.Result of Intraday and Interday precision of $\operatorname{PQP}(750 \mu \mathrm{~g} / \mathrm{ml})$

| Conc. $\boldsymbol{\mu g} / \mathbf{m l}$ | Time (hr.) | Mean Peak Area <br> $\mathbf{n}=\mathbf{6}$ | \%RSD | Time <br> (days) | Mean Peak Area <br> $\mathbf{n}=\mathbf{6}$ | \% RSD |
| :--- | :---: | :--- | :--- | :--- | :--- | :--- |
| $750(\mathrm{PQP})$ | 0 | 8569654 | $0.08 \%$ | 1 | 8569654 | $0.12 \%$ |
| $750(\mathrm{PQP})$ | 2 | 8569364 | $0.08 \%$ | 2 | 8559364 | $0.07 \%$ |
| $750(\mathrm{PQP})$ | 4 | 8568965 | $0.11 \%$ | 3 | 8558965 | $0.19 \%$ |

Table.6.Result of Intraday and Interday precision of AM ( $150 \mu \mathrm{~g} / \mathrm{ml}$ )

| Conc. $\boldsymbol{\mu g} / \mathbf{m l}$ | Time (hr.) | Mean Peak Area <br> $\mathbf{n}=\mathbf{6}$ | \%RSD | Time <br> (days) | Mean Peak Area <br> $\mathbf{n = 6}$ | \% RSD |
| :--- | :---: | :--- | :--- | :--- | :--- | :--- |
| $150(\mathrm{AM})$ | 0 | 1132502 | $0.17 \%$ | 1 | 1132512 | $0.16 \%$ |
| $150(\mathrm{AM})$ | 2 | 1132546 | $0.11 \%$ | 2 | 1102296 | $0.11 \%$ |
| $150(\mathrm{AM})$ | 4 | 1129956 | $0.17 \%$ | 3 | 1128250 | $0.17 \%$ |

System suitability tests were carried out on freshly prepared standard solution and all the parameters are within limit. Results were shown in table No:7

Table.7. System suitability data

| Parameters | $\mathbf{P Q P}( \pm \%$ RSD $)$ | $\mathbf{A M}( \pm \% \mathrm{RSD})$ |
| :--- | :--- | :--- |
| Retention Time | $3.39 \pm 0.13$ | $5.53 \pm 0.14$ |
| Theoretical plate | $4035 \pm 0.12$ | $3025 \pm 0.34$ |
| Tailing Factor | $1.05 \pm 0.12$ | $1.16 \pm 0.23$ |

$\pm \% \mathrm{RSD}=$ Percentage Relative Standard Deviation.

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The method was linear in the concentration range of $375-1125 \mu \mathrm{~g} / \mathrm{ml}$ for PQP and $75-225 \mu \mathrm{~g} / \mathrm{ml}$ for AM, with a correlation coefficient of 0.999 for PQP and 0.999 for AM.


Figure.11.Linearity plot of PQP


Figure.12.Linearity plot of AM

The study of accuracy of the developed method has been done. The recovery was found in the range of $101.2 \%$ for PQP and $100.7 \%$ for AM shown in Table. 8 , indicating the accuracy of method and the $\%$ RSD of PQP and AM is 0.50 and 0.50 respectively.

Table.8.Accuracy Data Of The Analysis Of Pqp And Am

| Concentration <br> Of Spiked level <br> $\boldsymbol{\%}$ | Amount Std added <br> $\boldsymbol{\mu g} / \mathbf{m l}$ |  |  | Total amount <br> $\mathbf{f o u n d} \boldsymbol{\mu g} / \mathbf{m l}$ |  | \% Recovery <br> $\boldsymbol{\mu g} / \mathbf{m l}$ |  | Mean |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :---: |
|  | PQP | AM | PQP | AM | PQP | AM | PQP | AM |  |
| 50 | 375.2 | 75.1 | 374.9 | 74.8 | 101.4 | 101.3 |  |  |  |
| 100 | 750.3 | 150.2 | 749.8 | 149.9 | 101.4 | 101.3 | 101.2 | 101.7 |  |
| 150 | 1125.2 | 225.1 | 1124.9 | 224.7 | 101.3 | 100.7 |  |  |  |

Table No: 9 Result of Peak Purity of Drug in Acid Stressed Degradation

| Name of Drug | Peak Purity Result | Conclusion |  |
| :--- | :--- | :--- | :--- |
|  | Purity Angle | Purity Threshold | Passed |
| Piperaquine phosphate | 0.048 | 0.214 | Passed |
| Arterolane maleate | 0.217 | 0.232 |  |

Table.10.Result of Peak Purity of Drug in Alkali Stressed Degradation

| Name of Drug | Peak Purity Result |  |  |
| :--- | :--- | :--- | :--- |
|  | Purity Angle | Purity Threshold | Conclusion |
| Piperaquine phosphate | 0.049 | 0.230 | Passed |
| Arterolane maleate | 0.063 | 0.234 | Passed |

Table.11.Result of Peak Purity of Drug in Peroxide Stressed Degradation

| Name of Drug | Peak Purity Result |  |  |
| :--- | :--- | :--- | :--- |
|  | Purity Angle | Purity Threshold | Conclusion |
| Piperaquine phosphate | 0.057 | 0.209 | Passed |
| Arterolane maleate | 0.120 | 0.233 | Passed |

Forced degredation were carried out on prepared sample solution and all the parameters are within limit i.e10-30\% of degredation and purity angle of the peak purity should be less than the purity threshold. Results were shown in table No: 12

Table.12.Result of Stress Study for PQP

| Standard Area | 8568156 | 8571651 | 8566376 | 8563546 |
| :--- | :--- | :--- | :--- | :--- |
| 8563656 | 8566677 |  |  |  |
| Average Area | $0.3 \%$ |  |  |  |
| \%RSD | Unstressed | Acid Stress | Alkali Stress | Peroxide Stress |
| Test No. | 1368.90 |  |  |  |
| Avg. Wt (mg) | 1369.12 | 1367.90 | 1368.5 | 1368.1 |
| Wt. taken (mg) | 8533457 | 7748379 | 7671577 | 6212356 |
| Area (Inj.1) | 8536557 | 7745630 | 7670521 | 6218563 |
| Area (Inj.2) | 8561457 | 7746869 | 7671528 | 6215369 |
| Avg. Area | 0.1 | 0.6 | 0.7 |  |
| \% RSD | 751.20 | 682.08 | 675.32 | 546.87 |
| Assay(mg/tab) | 100.8 | 90.8 | 89.9 | 72.8 |
| Assay (\%) | NA | 9.9 | 10.8 | 27.8 |
| \% Degradation |  |  |  |  |

Table.13.Result of Stress Study for AM

| Standard Area | 1126325 | 1125325 | 1125625 | 1125865 |
| :--- | :--- | :--- | :--- | :--- |
| 1125632 | 1125754 |  |  |  |
| Avg area | $0.2 \%$ |  |  |  |
| \%RSD | Unstressed | Acid Stress | Alkali Stress | Peroxide Stress |
| Test No. | 1368.92 |  |  |  |
| Avg. Wt. (mg) | 1368.10 | 1368.90 | 1368.5 | 1368.4 |
| Wt.taken(mg) | 1132632 | 1055269 | 1062062 | 1024698 |
| Area (Inj.1) | 1131642 | 1056325 | 1063256 | 1025636 |
| Area (Inj.2) | 1132263 | 055865 | 1062965 | 0.7 |
| Avg. Area | 0.1 | 0.5 | 134.67 |  |
| \% RSD | 150.00 | 93.6 | 90.5 |  |
| Assay ( mg/tab) | 6.7 | 139.58 | 10.2 |  |
| Assay (\%) | 100.8 | 7.5 |  |  |
| Degradation(\%) | NA |  |  |  |

## 4. CONCLUSION

The Proposed RP-HPLC method is suitable for simultaneous determination of PQP and AM in Combined dosage form without any interferences from each other. The accuracy of the methods was assessed by recovery studies at three different levels. The method was found to be precise as indicated by the repeatability analysis, showing \% RSD less than 2. All the parameters for both the drugs met the criteria of ICH guidelines for method validation. The developed method may be recommended for routine and QC analysis of the investigated drugs to provide simple, accurate and reproducible quantitative analysis for the determination of determination of PQP and AM in combined formulation.

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